1. Let us assume that a hydrogen atom is in a 3p state. Show that the radial part of its wave function is

$$u_{31}(r) = \frac{4}{81\sqrt{6}} e^{-\frac{r}{3}} r^2 (6-r).$$

- 2. Derive the dipole selection rules $\Delta l = \pm 1$, $\Delta m = \pm 1$ by calculating the matrix elements of the x and y components of the dipole operator $\mathbf{D} = e\mathbf{r}$. You may need the identity $(2\ell+1)\sqrt{1-t^2}P_\ell^{m-1}(t) = P_{\ell+1}^m(t) P_{\ell-1}^m(t)$.
- 3. Consider a one-dimensional anharmonic oscillator. Let the perturbation be of the form $H_1 = \sigma \hbar \omega x^3$. Calculate the matrix elements

$$\langle \psi_{n+3} | x^3 | \psi_n \rangle = \sqrt{\frac{(n+3)(n+2)(n+1)}{8}},$$

$$\langle \psi_{n+1} | x^3 | \psi_n \rangle = 3\left(\frac{n+1}{2}\right)^{\frac{3}{2}},$$

$$\langle \psi_{n-1} | x^3 | \psi_n \rangle = 3\left(\frac{n}{2}\right)^{\frac{3}{2}},$$

$$\langle \psi_{n-3} | x^3 | \psi_n \rangle = \sqrt{\frac{n(n-1)(n-2)}{8}}.$$

4. Consider an anharmonic oscillator in which the perturbation is of the form $H_1 = \sigma \hbar \omega x^3$. First, show that the energy levels are (to the second order in σ)

$$E_n = (n + \frac{1}{2})\hbar\omega - \sigma^2 \left[\frac{15}{4} (n + \frac{1}{2})^2 + \frac{7}{16} \right] \hbar\omega.$$

Then, show that the energy difference

$$E_n - E_{n-1} = \left(1 - \frac{15}{2}\sigma^2 n\right)\hbar\omega.$$

Finally, determine the energy eigenstates to the first order in σ . See the lecture notes section 17.2.3.

5. Calculate the van der Waals constant

$$C = \frac{e^4}{2E_{1s}} \langle 1s; 1s | (x_A x_B + y_A y_B - 2z_A z_B)^2 | 1s; 1s \rangle$$

$$= \frac{e^4}{2E_{1s}} 6 \left| \langle 1s | \frac{r_A^2}{3} | 1s \rangle \right|^2$$

$$= 6e^2 a_0^5.$$